"APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859430003-8

84659

Growth of Metal Monocrystals Under High Hydrostatic Pressure

s/020/60/135/001/011/030 B006/B056

The crystal structure was subjected to X-ray examination; the results obtained by these examinations are intended to be published in a later paper. There are 2 figures and 20 references: 12 Soviet, 2 German, 4 US, and 2 British.

ASSOCIATION:

Institut fiziki vysokikh davleniy Akademii nauk SSSR (Institute of Physics of High Pressures of the Academy

of Sciences USSR)

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SUBMITTED:

July 7, 1960

Card 3/3

s/020/60/135/005/018/043 во19/во67

AUTHORS:

Verenov, F. F., Vereshchagin, L. F., Corresponding Member of the AS USSR, and Goncharova, V. A.

TITLE:

Effect of Hydrostatic Pressure on the Elastic Properties

of Cerium

PERIODICAL:

Doklady Akademii nauk SSSR, 1960, Vol. 135, No. 5,

pp, 1104-1107

TEXT: In the introduction, the authors describe the cerium anomalies at high pressures and low temperatures which have been known for a considerable period of time. They mention A. I. Likhter, Yu. N. Ryabinin, L. F. Vereshchagin (Ref. 1), and Bridgman (Ref. 2). The authors studied the elastic properties at high pressure, and their changes on a polymorphous transition of the structure by means of ultrasonic methods. The propagation of ultrasonic waves with a frequency of 3.5 - 5.5 megacycles was measured by means of a pulse device shown in Fig. 1. Specimens contained 98.5% Ce, 1.5% of rare earths, 0.002% Fe, and 0.00% Pb, Bi, Sn, and Sb, their diameter was 20 mm, their lengths differed. With increasing

Card 1/4

Effect of Hydrostatic Pressure on the Elastic Properties of Cerium

S/020/60/135/005/018/043 B019/B067

pressure, phase transition took place at 7650 ± 50 kg/cm², with decreasing pressure, at 5950 ± 50 kg/cm². Formulas are given for calculating the elasticity characteristics from the results of sound velocity measurements. Figs. 2 and 3 graphically show the dependence of the propagation velocity of longitudinal waves, of the propagation velocity v_t of transverse v₁ of longitudinal waves, of the propagation velocity v_t of transverse waves, and the Debye temperature on hydrostatic pressure, as well as the dependence of elastic properties on hydrostatic pressure. There are 3 dependence and 15 references: 7 Soviet, 1 German, and 6 US.

ASSOCIATION: Institut fiziki vysokikh davleniy Akademii nauk SSSR (Institute of the Physics of High Pressures of the Academy of Sciences USSR)

SUBMITTED: August 22, 1960

Legend to Fig. 1: 1) trigger block, 2) pulse generator, 3) amplifier,

4) oscilloscope, 5) obtuator, 6) screwed nut, 7) high-pressure container,

8) electric supply lines, 9) piezoelement, 10) cerium specimens,

11) piezoreceiver.

Card 2/4

S/120/61/000/003/025/041 E194/E155

AUTHORS:

Bilevich, A.V., Vereshchagin, L.F., and

Kalashnikov, Ya.A.

TITLE:

A piezometer for determining the density of gases at

high pressures and temperatures

PERIODICAL: Pribory i tekhnika eksperimenta, 1961, No.3, pp.146-150

This article describes equipment which can be used to measure the compressibility of gases at pressures up to 3500 kg/cm2 at temperatures up to 400 °C with a total error not exceeding 0.1%. The novel features of the equipment are the high-pressure piezometer and miniature needle valve. A piezometer described by M. Benedict (Ref. 1: J. Amer. Chem. Soc., 1937, Vol.59, 2224) suffers from a number of practical disadvantages from which the present equipment is free. The main parts of the present author's piezometer are a thick-walled bulb 90 mm long, 8 mm internal diameter and 16 mm external diameter. It screws on to a head which carries a capillary tube with a highpressure needle valve. The needle valve, illustrated in Fig. 2, has a steel needle 1, a sealing nut 2 and a gland consisting of Card 1/4/

S/120/61/000/003/025/041 E194/E155

A piezometer for determining the ... E194/E155

three metal rings, one of copper 3, and two of steel 4, needle is ground to fit the inner diameter of the gland. The shank at the head of the needle is threaded to fit the threaded internal diameter of the sealing nut. The outer surfaces of the gland rings are a ground fit in the casing. Tests made with nitrogen at a pressure of 4.2 tons/cm2 and at room temperature, and at 3.5 tons/cm² and temperature of 400 °C, gave satisfactory results. Still higher values could no doubt be obtained if other grades of heat-resisting steel were used in the construction. The volume of the piezometer is about 5 me; it was carefully calibrated with carbon tetrachloride. In carrying out tests the piezometer is contained in a hollow copper block which is within a 300 W heating furnace. For purposes of weighing, the piezometer is suspended by a wire from the arm of an analytical balance which is on a bench above the furnace. The piezometer can thus be weighed without withdrawing it from the furnace. The arrangements that are made to fill the piezometer with clean gas and to measure the pressure on a standard manometer call for no comment. The following formula is used to calculate the change in volume Card 2/5

A piezometer for determining the ... \$\frac{\$\\$5/120/61/000/003/025/041}{\$\\$2194/\$\\$2155}\$

of the piezometer due to thermal expansion;

 $v_t = v_0(1 + 3.25 \times 10^5 t + 2.85 \times 10^{-8} t^2 - 1.65 \times 10^{-11} t^3)$

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An expression is also given for the change in volume due to pressure but when this was checked experimentally by a procedure which is described it was found to be in error. This can be seen from the curve of Fig.5, where the volume change as a function of pressure at temperatures of 21, 90 and 147 °C is plotted in tons/cm² as curve a. Curve b corresponds to the formula used, which is evidently inaccurate. The test procedure is as follows. The piezometer is heated to the test temperature, then filled with compressed gas and allowed to stand connected to the gas supply with the valve open for 20-30 minutes to equalise the pressure and temperature. The piezometer is then disconnected from the high-pressure gas supply with the needle valve closed and is weighed. The gas is then released and it is weighed again. The volume and weight of gas being accurately known under the given conditions of temperature and pressure, the density and other characteristics can be calculated.

Card 3/5

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A piezometer for determining the \$/120/61/000/003/025/041 E194/E155

B.K. Muratovskiy is thanked for his assistance. There are 5 figures and 7 references: 3 Soviet and the following 4 English Language references:

Rof. 1: as in text above.

Ref. 4: A.E.H. Love, Math. Theory of Elas., 1927, London.

Kef.5: P.W. Bridgman, J. Amer. Chem. Soc., 1937, Vol.59, 2233.

Ref. 6: P.W. Bridgman, Proc. Amer. Acad. Arts and Sci., 1935, Vol. 70, 1.

ASSOCIATION: Institut fiziki vysokikh davleniy, AN SSSR (Institute of High-Pressure Physics, AS USSR)

SUBMITTED: July 12, 1960

Card 4/5

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859430003-8"

S/120/61/000/002/041/042 E113/E135

AUTHORS: Stepanov, V.A., and Vereshchagin, L.F.

TITLE: High temperature resistance heater with graphite spiral heating element for high pressure vessels

PERIODICAL: Pribory i tekhnika eksperimenta, 1961, No. 2, pp. 194-195

TEXT: The design of the heater for heating the inside of vessels containing high pressure gas is shown in Fig.1. The heating element is a spiral (3), turned from a graphite rod together with robust end pieces. To increase the electrical contact area threads are cut in the end pieces on which the stainless steel contacts 1 and 10 are screwed. The heating element is surrounded by graphite tube (4) which is electrically insulated from the spiral by a pyrophyllite bush (6), put on the cylindrical neck of the intruding end piece. On each end the bush is fixed to the heating element by three porcelain pins (7), so that there is no relative movement between the spiral and the end pieces. The pins are in the relatively large uncut parts of the end pieces the end pieces have small ohmic resistance and are placed sufficiently far away from the spiral so that the pins cannot Card 1/43

5/120/61/000/002/041/042

High temperature resistance heater.. E113/E135

Around the graphite tube (4), five coiled molybdenum overheat. sheet shields, each 0.2 mm thick, are fitted concentrically. shields are spaced with gaps of 0.1 mm between them, which are maintained by means of two rows of pointed projections along the edges of the shields. The shields are held together by toothed rings; these create a gap of 2 mm between the inner wall of cover and the outer shield allowing only point contact at the teeth for Into this gap fireproof material in powder form heat conduction. is poured for additional heat insulation (for instance MgO). This gap and the annuli between the shields are closed with pyrophyllite washers (2) and (9). The length of the assembled heater is 120 mm, diameter 30 mm, corresponding to the diameter of the operating pipe of the high pressure vessel. The heater can be easily withdrawn and placed into the high pressure vessel by means of a rod screwed on the threaded part of contact (1)(Fig. 1). Contact (10) sits on the "finger" of the lead which is placed in the middle of the cover closing the high pressure vessel. other end of the spiral is connected to the wall of the pipe by means of contact (1) and in this way the electrical circuit is closed through the body of the vessel. The maximum working Card 2/4

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High temperature resistance heater.. E113/E135

temperature of the heater depends on the material used. Graphite has been chosen as it has the best properties for this application. A heater of given dimensions gave a temperature rise up to 2200 °C at 15 katm. pressure, with a power consumption of 2.8 kW to 3 kW. The effectiveness of the present method of heat insulation has been tested experimentally (in Ar, N_2 and CO_2 atmospheres), and it was found that at 2200 °C inside the heater and at 15 katm. after one hour of continuous operation, at a point in the wall of the high pressure vessel 15 mm from its inner wall, the temperature was 170 °C. The heater withstood satisfactorily several cycles of applying and relieving the pressure. Acknowledgements are expressed to I.Ye. Surkov and V.A. Frolov who assisted in the construction of the heater. (This is an abridged translation). There are 2 figures and 4 references: 3 Soviet and 1 English. ASSOCIATION: Institut fiziki bysokikh davleniy AN SSSR (Institute of High Pressure Physics, AS USSR)

SUBMITTED: January 29, 1960

Card 3/4

1145, 12 1045 also 108 s/126/61/011/002/005/025 18 9200 24 2130 10 55, 1000, 1164 E021/E435

Panova, G.Kh., Sekoyan, S.S. and Vereshchagin, L.F. Phase Diagram of Bismuth at Pressures and Temperatures AUTHORS:

up to 100000 kg/cm2 and 500°C TITLE :

PERIODICAL: Fizika metallov i metallovedeniye, 1961, Vol.11, No.2,

pp.215-219

The p-T phase diagram for bismuth had been investigated up to 100000 kg/cm² and 500°C in order to compare the results with The pressure equipment will be described in a A bismuth wire, 0.5 mm diameter, was placed in a other authors. The medium for transmitting the pressure was silver chloride which gives a quasi-hydrostatic pressure up to high The sample was heated by an electric current. pressure in the container was determined from the force developed The apparatus was calibrated from the known polymorphic transformations of bismuth (24800 and 27000 kg/cm2) thallium (43400 kg/cm²) and barium (77400 kg/cm²). The temperature was determined by the integral electrical power received by the wire after establishing that, with constant geometry of the sample

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5/126/61/011/002/005/025 E021/E435

Phase Diagram of ...

Card 2/5

and constant thermal conductivity of the surrounding medium, the temperature of the middle of the sample was linearly proportional to the power and practically did not change with change in specific heat conductivity of the investigated sample or with pressure. This was done using different metals at various pressures. method gave a temperature measurement with an accuracy of + 5 - 10°C and eliminated the disadvantage of using electrical leads required for other methods of measurement. In the investigations of the phase diagrams of bismuth, polymorphic transformations were detected by means of the rapid changes in the electrical resistance of the sample. The relation between the resistance R (ohms) and the power W (watts) received by the sample was established and Fig. 4 shows some of the results (xcubkccmb - liquid; curves 1 to 12 relate to pressures of 28000 to 100400 kg/cm2). From the results a phase diagram was constructed and is given in Fig. 5 (dotted line - data of Bundy; top left of diagram - "liquid"). The average accuracy of the results was estimated as 2% for both temperature and pressure. The results are in good agreement with those of F.P.Bundy (Ref.6).

\$/126/61/011/002/005/025 E021/E435

Phase Diagram of ...

There are 5 figures, 1 table and 9 references: 4 Soviet and

5 non-Soviet.

ASSOCIATION: Institut fiziki vysokikh davleniy

(Institute of Physics of High Pressures)

SUBMITTED: May 20, 1960

Card 3/5

1418,1138 18.8200

s/126/61/011/003/010/017 E032/E514

AUTHORS: Voronov, F.F. and Vereshchagin, L.F.

Effect of Hydrostatic Pressure on the Elastic Properties of Metals. 1. Experimental Data TITLE:

PERIODICAL: Fizika metallov i metallovedeniye, 1961, Vol.11, No.3, pp. 443-450.

The effect of hydrostatic pressure on the elastic properties of polycrystalline specimens of pure metals has been investigated. The mechanical properties were investigated using a pulse ultrasonic method (10 Mc/s). The ultrasonic apparatus employed was described earlier by the authors and V. I. Murav'yev The following materials were investigated: Al, 99.996% pure (Fe 0.0015%, Si 0.0015%, Cu 0.001%); Mg, 99.92% rure (Fe 0.04%, Si 0.01%, Cu 0.01%, Al 0.02%); armco-iron (Fe 99.8%, C 0.012%, Si 0.02%, Mn 0.02%, P.0.03%, S 0.03%); Be, 99.2% pure (Fe 0.36%, Mg 0.2%, Al 0.05%, Si 0.05%, Mn 0.02% Ni 0.015%; molybdenum, 99.88% pure (W 0.1%, Fe 0.005%, Al 0.002%, Cu, Zn, P, The densities of these materials were found to be: A1 - 2.695, magnesium - 1.731, armco-iron - 7.836, beryllium-1.843, molybdenum - 9.838 g/cm³. Quartz plat s, Mn, As 0.001%). Quartz plates were Card 1/#

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Effect of Hydrostatic Pressure ...

5/126/61/011/003/010/017 E032/E514

attached to the ends of the specimens and their working frequency was 10 Mc/s. The elastic moduli were computed from the following data: length of specimen under pressure (C), density (P), time of transit of longitudinal and transverse altrasonic waves at atmospheric pressure and the change in these times on application of pressure, Δt_{ℓ} and Δt_{t} . Changes in length and density on application of pressure were corrected for using the results of Bridgman and others. The adiabatic bulk modulus was calculated (1)from the formula

 $K_{t,0} = 4l_0^2 \rho_0 \left(l_{t,0}^{-2} - \frac{4}{3} l_{t,0}^{-2} \right)$

and the corresponding isothermal bulk modulus from

$$K_{T,0} = \frac{K_{g,0}}{1 + \frac{a^2 T K_{g,0}}{I \cdot C_g}},$$
 (2)

where α is the volume expansion coefficient, assumed to have the Card 2/10.

CIA-RDP86-00513R001859430003-8

Effect of Hydrostatic Pressure... S/126/61/011/003/019/017 E032/E514

following values for Al, Mg, Fe, Be and Mo, respectively: 7.09×10^{-5} , 7.66×10^{-5} , 3.52×10^{-5} , 3.65×10^{-5} , 1.54×10^{-5} deg. The specific heat C at constant pressure for these materials was assumed to be 0.214, 0.235, 0.180, 0.475, 0.060 cal/g·deg, respectively. The temperature was $T = 303^{\circ}K$, the mechanical equivalent of heat was assumed to be 4.182×10^{7} erg/cal and at 2000 kg/cm^2 the adiabatic bulk modulus was calculated (on the first approximation) from the formula

$$K'_{t,2} = 4l_0^2 \rho_0 \left(1 + \frac{1}{3} \frac{\Delta P}{K\tau_{t,0}}\right) \left[\left(l_{t,0} + \Delta t_i\right)^{-2} - \frac{4}{3} \left(l_{t,0} + \Delta t_i\right)^{-2} \right]. \tag{3}$$

The adiabatic bulk modulus $K_{5,2}^{\prime}$ was then converted to the isothermal bulk modulus $K_{T,2}^{\prime}$ in accordance with Eq.(2). Changes in C and α were neglected. Next, in order to introduce the correction, the average value of the bulk modulus in the range of to 2000 kg/cm was used and the second approximation $K_{5,2}^{\prime\prime}$ was computed. The new average value of the isothermal modulus was Card 3/10

Effect of Hydrostatic Pressure...

S/126/61/011/003/010/017 E032/E514

then used in the correction term for the next approximation. This procedure can be continued indefinitely to obtain the values of K Young's modulus, the shear modulus and the velocity of propagation of ultrasonic waves were computed from the formulae

$$E = \frac{3K_{I}}{A-1}, \text{ rise } (A = \frac{v_{I}^{2}}{v_{I}^{2}} = \frac{(I_{I,0} + \Delta I_{I})^{2}}{(I_{I,0} + \Delta I_{I})^{2}};$$
 (5)

$$G = 4l_0^2 \rho_0 \left(1 + \frac{1}{3} \frac{\Delta P}{K_{T,2}}\right) \left(1 + \frac{1}{3} \frac{\Delta P}{K_{T,4}}\right) \dots (l_{p,q} + \Delta l_q)^{-2}; \tag{6}$$

$$v = 2l_{\bullet} \left(1 - \frac{1}{3} \frac{\Delta P}{\overline{K_{T,2}}}\right) \left(1 - \frac{1}{3} \frac{\Delta P}{\overline{K_{T,4}}}\right) \dots (l_{\bullet} + \Delta l)^{-1}. \tag{7}$$

$$\sigma = \frac{1}{2} \frac{A - 2}{A - 1} \tag{8}$$

Card 4/10

 Effect of Hydrostatic Pressure... S/126/61/011/003/010/017 E032/E514

The Debye temperature was calculated from

$$\Theta_{P} = \frac{h}{h} \left(\frac{9 N_{A}}{4 \pi} \right)^{\frac{1}{3}} l_{0} \left(\frac{\rho_{0}}{M} \right)^{\frac{1}{3}} \left[(l_{1,0} + \Delta l_{i})^{3} + 2 (l_{1,0} + \Delta l_{i})^{3} \right]^{-\frac{1}{3}}, \tag{11}$$

and the average velocity of sound from

$$\overline{v} = \left(\frac{1}{v_i^3} + \frac{2}{v_i^3}\right)^{-\frac{1}{3}} \tag{10}$$

In the above expressions N_a is the Avogadro number. All the results were obtained at 30°C. The numerical data are summarized in Figs. 1-5, in which the pressure is plotted along the horizonal axes in kg/cm². There are 5 figures, 2 tables and 16 references: 3 Soviet and 13 non-Soviet.

ASSOCIATION: Institut fiziki vysokikh davleniy AN SSSR (Institute of Physics of High Pressures AS USSR)

SUBMITTED: July 22, 1960

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1. 1210

AUTHORS:

Vereshchagin, L. F., Corresponding Member of the AS USSR,

Semerchan, A. A., Kuzin, N. N., and Popova, S. V.

TITLE:

Changes in Resistivity of Some Metals at Pressures of up

to 200 000 kg/cm²

Doklady Akademii nauk SSSR, 1961, Vol. 136, No. 2, pp. 320-321 PERIODICAL:

TEXT: The authors studied the resistivity of antimony, arsenic, and calcium at pressures of up to 200 coo kg/cm^2 . Likewise, bismuth, whose resistivity has hitherto been known up to 140 000 kg/cm^2 , was investigated. The bismuth and calcium specimens were made from wire, the antimony and arsenic specimens were thin single crystals. All specimens were chemically pure. As may be seen from changes in resistivity of the specimens graphically represented in Figs. 1, 2, and 3, arsenic and calcium have a monotonic change of resistivity with rising pressure, bismuth and antimony, however, have not. At 130 000 kg/cm^2 , antimony shows a jump-like change

Card 1/6 2

Changes in Resistivity of Some Metals at Pressures of up to 200 000 kg/cm²

s/020/61/136/002/012/034 B019/B056

in resistivity, bismuth at 125 000 kg/cm². The authors point out the possible use of the jump-like change in resistivity of antimony at 130 000 kg/cm² for the calibration of high-pressure devices. A parallel connection of antimony and bismuth (Fig. 18) would be part:cularly suited. There are 4 figures and 2 references: 2 US.

ASSOCIATION:

Institut fiziki vysokykh davleniy Akademii nauk SSSR (Institute of the Physics of High Pressures of the Academy of Sciences USSR)

THE TYPE:

October 10, 1960

Card 2/6

s/020/61/138/001/011/023 B104/B201

F., Corresponding Member of the AS USSR,

Semerchan, A. A., Kuzin, N. N., and Popova, S. V.

TITLE:

Change of resistivity of some metals at pressures up to

250,000 kg/cm²

PERIODICAL:

Doklady Akademii nauk SSSR, v. 138, no. 1, 1961, 84-85

TEXT: This is in continuation of an earlier paper by Vereshchagin et al. (DAN, 136, no. 2, (1961)). The authors wanted to find new polymorphous transformations at high pressures in metals being accompanied by an abrupt change of resistivity. Bridgman (Proc. Am. Acad. Arts and Sci., 81, 165 (1952)) and Bundy (Phys. Rev., 110, no. 2, (1958)) have been able to identify a considerable number of polymorphous transformations of various metals and alloys at high pressures. The possibility is pointed out of calibrating high-pressure apparatus with the aid of an abrup; change of the resistivity of different alloys at given pressures. The authors used a high-pressure chamber calibrated with the aid of the known resistivity

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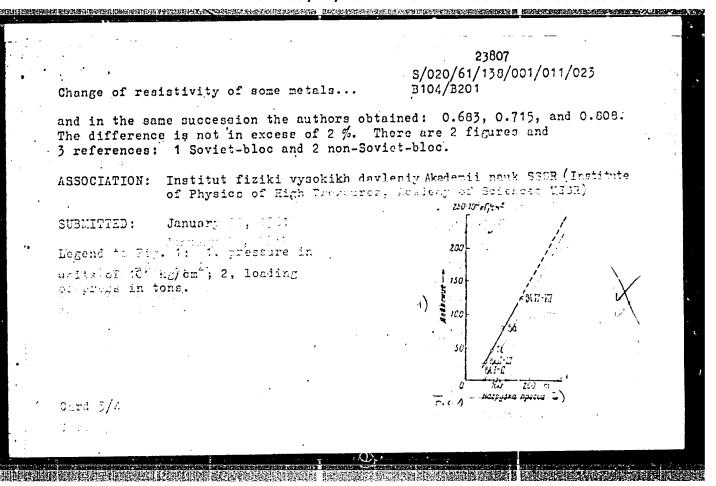
Change of resistivity of some metals...

jumps to determine the resistivity of the following metals: Bi I-II $(25,600 \text{ kg/cm}^2)$; Bi II-III $(27,000 \text{ kg/cm}^2)$; T1 $(45,000 \text{ kg/cm}^2)$; Ba $(80,000 \text{ kg/cm}^2)$; Bi VI-VII $(125,000 \text{ kg/cm}^2)$. Pressure above $(125,000 \text{ kg/cm}^2)$ as determined by extrapolation $(125,000 \text{ kg/cm}^2)$. The specimens were wires $(125,000 \text{ kg/cm}^2)$ and in diameter, the medium transmitting the pressure was silver chloride. Measurements were conducted at room temperature. Measurement results are graphically presented in Fig. 2. $(125,000 \text{ kg/cm}^2)$ is the resistivity of the metal concerned at a pressure of $(125,000 \text{ kg/cm}^2)$. Pridgman discovered on zirconium at a pressure above $(125,000 \text{ kg/cm}^2)$. Bridgman discovered on zirconium at a pressure above $(125,000 \text{ kg/cm}^2)$. The authors have not been able to ascertain this drop up to $(125,000 \text{ kg/cm}^2)$. The difference in results is explained by a possible difference in the purity degree of the metals. The authors used zirconium iodide with $(125,000 \text{ kg/cm}^2)$. The following comparative data are offered: Bridgman obtained for Pb: $(125,000 \text{ kg/cm}^2)$.

 $R_{100}/R_{30} = 0.707$, for Cd: $R_{100}/R_{30} = 0.795$. Under the same conditions

Card 2/4

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s/020/61/138/305/009/025 B104/B205

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AUTHORS:

also 2108

Vereshchagin, L. F., Corresponding Member AS USSR,

Semerchan, A. A., and Popova, S. V.

TITLE:

Study of the electrical resistance of cerium, lanthanum, and

neodymium at pressures of up to 250,000 kg/cm2

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 138, no. 5, 1961, 1059-1061

TEXT: This is the continuation of previous papers by the authors (DAN, 136, no. 2, (1961); DAN, 138, no. 1 (1961)), in which the electrical resistance of metals at high pressures (up to 250,000 kg/cm2) has been studied systematically. P. " Bridgman (Proc Am. Acad Arts and Sci., 81, 165 (1952)) proved that cerium, lanthanum, and neodymium have a minimum at Cerium shows a minimum at pressures ranging from 50,000 to 100,000 kg/cm². Similar results were ob-70,000 kg/cm² and a maximum at 90,000 kg/cm². tained by Bridgman for the other two metals. The investigations described here were conducted with a high-pressure chamber which had been calibrated with the help of known sudden changes of the electrical resistance of cer-

Card 1/8,

的。 第一个时间,我们就是我们的,我们就是我们的,我们就是我们的,我们就是我们的,我们就是我们的,我们就是我们的,我们就是我们的,我们就是我们的,我们就是我们的一个人

> S/020/61/138/005/009/025 B104/B205

Study of the electrical resistance of ...

tain pressures. The results are graphically represented in Figs. 2-4 The change of the electrical resistance R/R_{30} (R_{30} is the electrical resistance at a pressure of 30.000 kg/cm²) shown in Fig. 2 indicates that cerium has a minimum at 55,000 kg/cm² and a maximum at 80,000 kg/cm². maximum of the electrical resistance is taken as an indication of a polymorphous conversion occurring at this pressure. Fig 3 shows analogous curves obtained for two specimens of lanthanum of varying purity: La-I (0.75% Nd, 0.70% Pr. 0.04% Fe) and La x 11 (0.3% Nd, U.2% Pr. 0.02% Fe). It may be seen that only the last-mentioned type of (chemically pure) lanthanum has a weakly marked minimum at a pressure of approximately 95,000 kg/cm² and weakly marked maxima at 110,000 and 140,000 kg/cm². It is assumed that a polymorphous conversion takes place also here at 110,000 kg/cm². Fig. 4 indicates that neodymium has indistinct minima and maxima at 80,000 and 90,000 kg/cm², respectively. This maximum is likewise ascribed to a polymorphous conversion. The different values of maxima and minima on the resistance curves are explained as being due to a great calibration error. All measurements were made with specimens in wire form

Card 2/6

 25309 S/020/61/138/005/009/025 Study of the electrical resistance of ... B104/B205

(1-1.5 mm diameter) at room temperature. Cerium impurities: less than 0.75% Nd, less than 0.75% Pr, $2\cdot10^{-2}\%$ Fe, $1\cdot10^{-3}\%$ Cd, $1\cdot10^{-3}\%$ Pb, $1\cdot10^{-3}\%$ H, and $1\cdot10^{-3}\%$ Sn; neodymium impurities: less than 0.36% Pr and La, and $2\cdot10^{-2}\%$ Ca. Following this series of articles, the authors will present a theoretical discussion of their results. There are 4 figures and 3 references: 2 Soviet-bloc and 1 non-Soviet bloc.

ASSOCIATION: Institut fiziki vysokykh davleniy Akademii nauk SSSR

(Institute of Physics of High Pressures of the Academy of

Sciences USSR)

SUBMITTED: March 4, 1961

Card 3/6

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Card 1/#2

Vereshchagin, L. F., Corresponding Member of the AS USSR,

25713

B104/B201

\$/020/61/139/003/012/025

Semerchan, A. A., and Popova, S. V. AUTHORS:

Change of electric resistance of praseodymium, dysprosium, erbium, and ytterbium at pressures of up to 250,000 kg/cm^2 TITLE:

Akademiya nauk SSSR. Doklady, v. 139, no. 3, 1961, 585 - 586 PERIODICAL:

TEXT: This is the fourth report on studies conducted on changes of electric resistance of metals at high pressures (Vereshchagin et al., DAN, 136, no.2, (1961); DAN, 138, no. 1. (1961); DAN. 135. no. 5. (1961). The change of relative resistance R/R₂₅ (R₂₅being resistance at a pressure of 25,000 kg/cm) of praseodymium is graphically shown in Fig. 1. Reference is made to the minimum appearing at about 110.000 kg/cm^2 , and it is stated that this pressure dependence of resistance is the same as the one in lanthanum; praseodymium and lanthanum exhibit the same crystal structure. In both of them, a polymorphous transformation of the crystal structure is believed to take place at this pressure According to measurements by

25713 \$/020/61/139/003/012/025 B104/B201

Change of electric resistance of

P. W. Bridgman (Proc. Am. Acad. Arts and Sci., 1952, 81, 165 (1952)) the minimum for praseodymium is at 80,000 kg/cm². This difference is explained by a different degree of purity of the material. The relative resistance of dysprosium as a function of pressure is shown in Fig. 2. There is a minimum at 75,000 kg/cm²; this dependence equals that of reodymium. Data for erbium are graphically presented in Fig. 3; for ytterbium, they are given in Fig. 4. The strongly pronounced maximum at 50,000 kg/cm² is explained by a polymorphous transformation or by an electron transition. Cerium exhibits the same dependence between relative resistance and pressure; both metals have a cubically face-centered lattice. In the following papers, the authors will examine the resistance of lanthanides as a function of pressure. There are 4 figures, 1 table, and 4 references: 3 Soviet-bloc and 1 non-Soviet-bloc.

ASSOCIATION:

Institut fiziki vysokikh davleniy Akademii nauk SSSR (Institute of High-pressure Physics, Academy of Sciences USSR)

Card 2/4

S/181/62/004/006/035/051 B108/B138

AUTHORS:

Frolov, A. P., Vereshchagin, L. F., and Rodionov, K. P.

TITLE:

Changes in the lattice parameters of pentaerythrite under

pressures of up to 10,000 kg/cm²

PERIODICAL:

Fizika tverdogo tela, v. 4, no. 6, 1962, 1608-1612

TEXT: A radiographic investigation of the lattice parameters a and c of pentaerythrite: $C(CH_2OH)_4$ was made in a beryllium high-pressure chamber. At room temperature with pressures up to 10,000 kg/cm², pentaerythrite has a tetragonal crystal lattice with the parameters a = 6.10 Å and c = 8.73 Å. At a pressure of 9,000 kg/cm², a = 5.99 Å and c = 8.46 Å. A discontinuity of a and c was observed between 4200 and 5600 kg/cm². The volume also

of a and c was observed between 4200 and 5600 kg/cm². The volume also of a and c was observed between 4200 and 5600 kg/cm². The volume also of a and c was observed between 4200 and 5600 kg/cm². The volume also of a phase changed suddenly by some 2.6%. These data are evidence of a phase changed suddenly by some 2.6%. These data are evidence of a phase transition in which, however, the crystal structure below and above the transition pressure remained the same. The behavior of pertaerythrite transition pressure can be described by two empirical third-order equations of under pressure can be described by two empirical third-order equations of state:

Card 1/2

S/181/62/001/006/035/051 B108/B138

Changes in the lattice ...

Before transition: $-\frac{\triangle V}{V_0} = 1.584 \cdot 10^{-5} P - 2.380 \cdot 10^{-9} P^2 + 0.330 \cdot 10^{-13} P^3;$ after transition: $-\frac{\triangle V}{V_0} = 2.404 \cdot 10^{-5} P - 3.848 \cdot 10^{-9} P^2 + 2.202 \cdot 10^{-13} P^3.$

Above the pressure of transition, compressibility increases with increasing pressure. There are 5 figures and 1 table.

Institut fiziki metallov AN SSSR, Sverdlovsk (Institute of ASSOCIATION:

Physics of Metals AS USSR, Sverdlovsk). Institut fiziki

vysokikh davleniy AN SSSR, Moskva (Institute of High-

pressure Physics AS USSR, Moscow)

SUBMITTED:

February 15, 1962

Card 2/2

S/181/62/004/007/030/037 B178/B104,

AUTEORS:

Yevdokimova, V. V., and Vereshchagin, L. F.

TITLE:

Polymorphous transition in NaCl

的现在,我们就是一个人的人,我们就是这些人的人,我们们的人们的人,我们们们的人们的人们是一个人的人们是一个人的人们的人们的人们是一个人们的人们的人们的人们,我们

PERIODICAL:

Fizika tverdogo tela, v. 4, no. 7, 1962, 1965-1966

TEXT: At pressures close to 1.8·10⁴ kg/cm², NaCl acquires a structure of the CsCl type. Its original lattice is stable, the constant equals 3.39-0.06 % at atmospheric pressure. The density of the new phase is 2.535 g/cm², and the discontinuity in the specific volume during the transformation is 14.2%. Allowing for the fact that the distance between the oppositely charged ions increases by 3% the lattice constant of the new phase is found to be 3.35 %. When the pressure is released, the new phase is usually maintained. Shear deformation might play a significant role in the phase transition. There are 1 figure and 1 table.

ASSOCIATION:

Institut fiziki vysokikh davleniy AN SSSR Moskva

(Institute of the Physics of Pressures AS USSR, Moscow)

SUBMITTED:

March 19, 1962

Card 1/1

FROLOV, A.P.; VERESHCHAGIN, L.F.; RODIONOV, K.P.; OLEYNIK, M.I.

Methods of X-ray investigation of materials under high pressures. Part 2: Equipment for the preparation of X-ray pictures of powders under pressure of up to 18,000 kmcm². Fiz. net. i metalloved. 14 no.1:80-84 Jl 162. (MCRA 15:7)

1. Institut fiziki metallov AN SSSR i Institut fiziki vysokikh davleniy AN SSSR.

(Metal powders) (X rays—Diffraction)

S/057/62/032/002/016/022 B124/B102

AUTHORS: Vereshchagin, L. F., Zubova, Ye. V., and Shapochkin, V. A

TITLE: Electric contact resistance at high normal pressures

PERIODICAL: Zhurnal tekhnicheskoy fiziki, v. 32, no. 2, 1962, 230 - 232

TEXT: The pressure dependence of the electric contact resistance of pistons made of UX15 (ShKh15) steel and of a powder-metallurgical hard alloy of the type \$K6 (VK6) was examined at pressures of up to 100,000 kg/cm², using the high-contact-pressure method developed at the authors institute. The purity and the quality of the contact surfaces were kept constant in all experiments. The diameter of the rated contact area of the pistons was also constant and equal to 3 or 6 mm. The electric contact was calculated from the change in contact resistance measured with a potentiometer of type MTH-1 (PPTN-1) and a high-sensitivity galvanometer of type M21/4 (M21/4) with low internal resistance. Heating of the contact and the relevant change in resistance were excluded by using 1- to 2-ma currents. The voltage drop was measured for two current directions, and the average value was determined. The contact resistance was calculated from Card 1/2

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S/057/62/032/002/016/022 B124/B102

Electric contact resistance ...

 $R_{\chi} = \frac{R_n U_{\chi}}{U_n}$, where R_n is the standard resistance, U_n is the voltage drop on

the standard sample, and $U_{\rm X}$ is the voltage drop on the sample examined. Pressure was gradually raised by 1,000 to 10,000 kg/cm² up to 100,000 kg/cm. Voltage drop measurements were repeated 15 to 20 times, and each test 3 to 4 times, with the first test results being neglected, as a rule. The results shown in Fig. 2 are in good agreement with those of other authors. There are 2 figures and 4 Soviet references.

ASSOCIATION: Institut fiziki vysokikh davleniy AN SSSR, Moskva (Institute of High-Pressure Physics, AS USSR, Moscow)

SUBMITTED: February 6, 1961

Card 2/3

3h215 3/057/62/032/002/017/022 B124/B102

15.2240 AUTHORS:

Vereshchagin, L. F., Shapochkin, V. A., and Pirogova, L. B.

TITLE

Contact compressive strength of hard alloys of type \$K(VK)

PERIODICAL: Zhurnal tekhnicheskoy fiziki, v. 32, no. 2, 1962, 235 - 237

。在1975年的19

TEXT: The compressive strength and residual properties (strength and plasticity) of pistons made of the sintered carbides BK8g(VK8V), BK6g(VK6V) BK6(VK6V), and BK4g(VK4V) were tested using special device. The diameter of the contact surface was 3-3.5 mm. Pistons were tested by applying only perpendicular pressure or perpendicular pressure and torque simultaneously. In the former case, the load was raised first to 100,000 kg/cm², then the sample was unloaded and examined for cracks, and loaded again at intervals

of 100,000 kg/cm² until the first cracks appeared. In the latter case stepwise loading by 10,000-20,000 kg/cm² was used, and, at a certain perpendicular pressure, torque was applied until the first cracks appeared. The results indicate that the contact compressive strength of the alloys increases with decreasing cobalt content. The breaking load is lowered by 20% when

Card 1/3

34215 3/057/62/032/002/017/022 B124/B102

Contact compressive strength

torque has been applied. The highest perpendicular load (350,000 kg/cm²) could be applied to VK4V pistons, while VK84 pistons cracked under a pressure of 220,000 kg/cm². Application of torque to VK4V gave most pronounced effects; the breaking load of the BK6 TaC(VK6TaS) alloy was about 200,000 kg/cm² It was found by microhardness tests with the device WAT-3 (PMC-3) that (1) microhardness increases equally both with perpendicular pressure and with pressure plus torque; (2) cold hardening of the contact surface is constant at all points of the surface except the border; (3) residual hardening reaches a maximum in VK4V (about 20%) and a minimum in VK6V (about 5%) Radial and annular cracks were formed in positions and distributions dependent on the kind of load. Tangential stresses as a function of perpendicu lar pressure were measured for VK8V. VK6V, and VK4V between 10,000 and 200,000 to 300,000 kg/cm2. The friction coefficients of all alloys at pressures up to 250,000 - 300,000 kg/cm² were all about 0.185, with a 1.5 to 2-fold decrease with increasing pressure. Mechanic L. M. Voyeykov and laboratory assistant Z. A. Levchenko are thanked. There are 5 figures. 1 table, and 5 Soviet references.

Card 2/3

34215 \$/057/62/0 32/002/017/022

Contact compressive strength ...

ASSOCIATION: Institut fiziki vysokikh davleniy AN SSSR, Moskva (Institute of High-pressure Physics, AS USSR, Moscow)

February 5, 1961 SUBMITTED:

Card 3/3

VERESHCHAGIN, L.F.; YUZEFOVICH, N.A.

Effect of pressure on the state of isomeric molecules. Zhur.
fiz.khim. 36 no.5:969-972 My '62. (MIRA 15:8)

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1. Institut fiziki vysokikh davleniy, AN SSSR.
(Alcohols) (Ultrasonic testing)

5/056/62/043/004/ 014/061 B102/B180

ANTHORS: Yevdoli ova, V. V., Vereshchagin, L. F.

Tibha Polymorphic transition in sodium chloride under pressure

PERICULOADA Zhurnal elsperimentalinoy i teoreticheskoy fiziki, v. 43, no. 4(10), 1962, 1208 - 1212

THAT: Pressures of up to 13,000 kg/cm² were applied to powder samples of "EMStra" table calt and of RaCl single crystals and the pressure-induced changes in volume and structure were studied by X-ray analysis. A new phase appeared at 17,700 kg/cm², i. e. the cubic face-centered lattice of phase appeared at 17,700 kg/cm², i. e. the cubic face-centered lattice of of 3.36±9.04Å. The density of the new phase is p = 2.535 g/cm³, that of the initial phase c = 2.165 g/cm³, so that the volume change $\Delta V = 14.2\%$ the initial phase c = 2.165 g/cm³, so that the volume change $\Delta V = 14.2\%$ the initial phase, 0.50b (a <0.63 b. The pressure dependence of the change in initial phase, 0.50b (a <0.63 b. The pressure dependence of the change in volume can be describe by -\Delta V/V = 14.30·10⁻² +36.0·10⁻⁷ p-60.0·10⁻¹² p. Gard 1/2

Polymorthic transition ...

3/056/62/045/004/014/061
2102/2100

There Voice the molecular volume of initial NaUl. After pressure relief about 2/2 of the new phase utill remained. There are 2 figures and 3 tables.

Association: Institute fiziki vysokikh davlenty Akademii nauk 335R (Institute of the shysics of liter Pressures of the Academy of Sciences 105h)

3-3-17139: May 15, 1962

Card 2/2

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以是此意识。我是明显是有非常是可能的思考的。对于这种,我们就是不是一种的,我们就是不是一个,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的
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                                           Kabalkina, S. S., and Vereshchagin, L. F., Corresponding
                                                                                                  B104/B102
                                             An X-ray diffraction study of the effects of hydrostatic
                                              All A-ray dillraction study of the effects of hydrostatic pressure up to 18,000 kg/cm<sup>2</sup> on the structure of lead titmate
                                            Member AS USSR
                                              Akademiya nauk SSSR. Doklady, v. 143, no. 4, 1962, 818 - 821
                      AUTHORS:
                         TEXT: The investigation was carried out at room temperatures by means of
                         TEXT: The investigation was carried out at room temperatures by means of a special high-pressure chamber (Fig. 2) with a fine-focus tube and copper the special high-pressure chamber (Adligated by the Fiziko-khimicheskiv
                       TITLE:
                          a special nign-pressure champer (rig. 2) with a line-locus tupe and consider the ceramic samples were delivered by the Fiziko-khimicheskiy and the ceramic samples (Physicochemical Tastitute im 1. Ye karoova (Physicochemical Tastitute im 1. Ye karoova (Physicochemical Tastitute im 1. Ye
                          PERIODICAL:
                           and had the following lattice parameters: a = 3.903 Å, c = 4.154 Å, the call tetragonality c/a = 1.064. With increasing pressure a decrease of the call tetragonality
                            and a lowering of the Curie point are observed. At 18,000 kg/cm<sup>2</sup> c is considerably smaller (Ac = -0.10 A) and a is slightly sreater, (As considerably smaller (Ac = -0.10 A) and a linear function of pressure p: +0.01 A). The relative change of c is a linear function of pressure p: -7
                              Ac/c = 14.3.10-7.p. The change of parameters with increasing pressure
                               Card 1/2
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5/020/62/143/004/010/027 An X-ray diffraction ... B104/B102 coincides qualitatively with their change as temperature functions. At 480° C $\Delta c = -0.129$ X and $\Delta a = 0.048$ X. High pressure and high temperature lower the polarization. The compressibility of the ferroelectric phase of PbTiO, is assumed to be a superposition of normal compression and deformation combined with a decrease of polarization under pressure. There are 4 figures and 1 table. SUBMITTED: December 29, 1961 Fig. 2. High-pressure chamber. Legend: (1) internal cylinder; (2) external cylinder; (3) sample; (4) beryllium cone; (5) lead-in, to which a manganin manometer is connected; (6) liquid; (7) diaphragm; (E) piston. Card 2/2

3/020/62/144/005/004/017 3125/3104

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Kabalkina, S. S., Vereshchagin, L. F., Corresponding AUTHORS:

Member AS USSR, and Shulenin, B. M.

TITLE:

X-ray study of the effect of hydrostatic pressure on the

structure of barium titanate

Akademiya nauk SSSR. Doklady, v. 144, no. 5, 1962, PERIODICAL:

1019-1021

TEXT: The effect of hydrostatic pressure on the structure of barium titanate was studied at room temperature. X-ray pictures with reflection

angles of 60-800 were recorded under pressures of 1-6000 kg/cm², using a KFOC (KROS) X-ray camera and an auxiliary high-pressure unit. The barium titanate specimens (lattice constants, a=3.993 Å and c=4.032 Å; Curie temperature T_{Cur} = 118°C) had been supplied by the Fiziko-

khimicheskiy institut im. L. Ya. Karpova (Physicochemical Institute imeni D. Marpov). The values of a, c, and Tour at high pressures were determined using the line group with $h^2+k^2+1^2=26$ (0 = 77-80°).

Cará 1/3

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S/020/62/141/005/004/017 B125/B104

X-ray study of the effect of ...

The X-ray pattern of BaTiO3 is shifted under high pressure but returns to its original position when the specimen is unloaded. $(1/p) \angle V/V$ was found to be 15.2.107 cm2/kg. As the pressure is increased from 1 to 6030 kg/cm², c decreases from 4.033 to 4.020 $^{\circ}$, a from 3.335 to 3.990 $^{\circ}$, and c/a from 1.010 to 1.0085, while $\triangle a/a$ increases from 0 to $\sim 0.13\%$, En _c/c from 0 to 0.25%. Most of these changes are linear in first convinction. Decrease of the Curie temperature diminishes the "intragonality" of the lattice. The pressure dependence of a, c, and in BaTiO, is qualitatively in accordance with the dependence of the respective quantities of the solid solution (Ba-Sr)TiO, on its content of SrPio, The compressibility Ac/c of PbTiO, is almost four times that of PaTiO2. The ferroelectric phase becomes compressible by the superposition of deformation and normal compression, attended by a decrease in polarization. The stretching of the PbTiO, lattice in the a-direction is of larger amount than the normal compression, and that of the $BaTiO_\chi$ Card 2/3

X-ray study of the effect of ...

s/020/62/144/005/004/017 B125/B104

lattice is smaller. The hydrostatic pressure affects the structure of lead titanate much more than that of barium titanate. There are A figures and 2 tables. The most important English-language reference is: 3. Merz, Phys. Rev., 78, 52 (1950).

ACCIATION: Institut fiziki vysokikh davleniy Akademii nauk SSSR (Institute of the Physics of High Pressures of the Academy

of Sciences USSR)

WENTTED:

February 23, 1962

Card 3/3

VERESHCHAGIN, L.F.; YUZEFOVICH, H.A.; CHELOVSKIY, A.V.

Measurement of ultrasound velocities in some highly compressed gases. Dokl. AN SSSR. 144 no.6:1272-1274 Je 162. (MIRA 15:6)

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 Institut fiziki vysokikh davleniy Akademii nauk SSR.
 Chlen-korrespondent Akademii nauk SSSR (for Vereshchagin). (Ultrasonic waves -- Speed) (Gases, Compressed)

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859430003-8"

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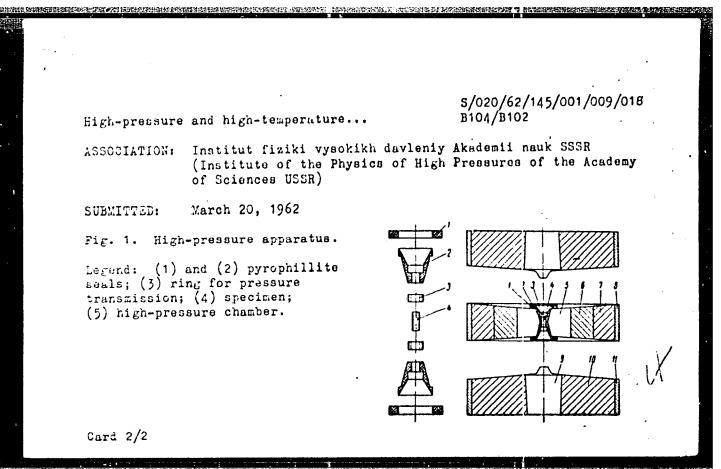
AUTHORS: Vereshchagin, L. F., Corresponding Member AS USSR, Semerchan, A. A., Zubkov, V. M., and Kuzin, N. N.

TITLE: High-pressure and high-temperature apparatus with several pairs of electric lead-in wires

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 145, no. 1, 1962, 71-72

TEXT: Difficulties arising in the current feed to high-pressure apparatus were overcome by the device shown in Fig. 1. Specimen 4 is placed in a cylindrical container inside a high-pressure chamber 5. Two pistons 9 compress the specimen. During compression the pyrophillite seals 2 enter the gaps (~ 0.1 mm) between the four sectors of pistons 9. The current is fed through the piston to the cylindrical graphite or metal container which is used as a furnace. The apparatus was calibrated for pressures of up to 50,000 kg/cm² by making use of the jumps known to occur in the electric conductivity of Bi and Tl at certain temperatures. There are 3 figures.

Card 1/2



VERESHCHAGIN, L.F.; SEMERCHAN, A.A.; POPOVA, S.V.; KUZIN, N.N.

Variations in the electric resistance of certain semiconductors at pressures up to 300,000 kg./cm.². Dokl.AN SSR 145 no.4:757-760 Ag 162. (MIRA 15:7)

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Ghlen-korrespondent AN SSSR (for Vereshchagin).

(Semiconductors-Electric properties)

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SEMERCHAN, A.A.; KUZIN, N.N.; VERESHCHAGIN, L.F.

Temperature dependence of the electric resistance of polycrystalline graphite at pressures up to 250,000 kg./cm². Dokl. AN SSSR 146 no.4:803-804 0 '62. (MIRA 15:11)

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Chlenkorrespondent AN SSSR (for Vereshchagin). (Graphite crystals—Electric properties) (High-pressure research)

KUZIN, N.N.; SEMERCHAN, A.A.; VERESHCHAGIN, L.F.; DROZDOVA, L.N.

Temperature dependence of the electroconductivity of iodine at pressures up to 200,000 kg./cm2. Dokl. AN SSSR 147 no.1:78-79 N '62. (MIRA 15:11)

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Chlen-korrespondent AN SSSR (for Vereshchagin).

(Iodine-Electric properties)

(High-pressure research)

ACCESSION NR: AP4009138

s/0056/63/045/006/2073/2076

AUTHORS: Kabalkina, S. S.; Vereshchagin, L. F.; Shulenin, B. M.

TITLE: Phase transitions in tellurium at high pressures

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SOURCE: Zhurnal eksper. i teoret. fiziki, v. 45, no. 6, 1963, 2073-2076

TOPIC TAGS: tellurium high pressure, phase transition, reversible phase transition, tellurium crystal structure, x ray diffraction pattern, x ray diffraction, Patterson Harker section, chain structure, laminar structure

ABSTRACT: An x-ray diffraction study of tellurium was carried out at pressures up to 100 kbar in order to find how the crystal structure of tellurium changes at high pressure. Two reversible phase transitions were observed, at 15--20 and 42--45 kbar. At 15 kbar tellurium is shown to undergo a transition from the chain structure

Card 1/2

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ACCESSION NR: AP4009138

A8 to the laminar structure A7, and the reasons why this transition was not detected by Bridgman (Proc. Am. Acad. Arts Sci. v. 60, 366, 1925) are explained. The structure of the second phase transition at 42--45 kbar could not be ascertained, but the constancy of the x-ray diffraction patterns above 45 kbar seems to cast doubts on the 69 kbar phase transition detected by Bridgman (Proc. Am. Acad. Arts Sci. v. 74, 425, 1942). Orig. art. has 2 figures and 2 tables.

ASSOCIATION: Institut fiziki vysokikh davleniy Akademii nauk SSSR (High Pressure Physics Institute, Academy of Sciences SSSR)

SUBMITTED: 11Sep63

DATE ACQ: 02Feb64

ENCL: 00

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NO REF SOV: 002

OTHER: 005

Card 2/2

20000- 10-1/ps-1 1-6-1 + 10/c	a dimentalist due j'ant jam
AUTHOR: Semeronan, A. A. Vires Kuzin, A. N., Drizdina, L. S.	ar mewown () P. — presidenting between AN വാർ എ. ജൂട്ട
PITIE: Changes in the resistivi pressures of up to 200,000 ke/gm	ity of PbTe, CdTe, and Bi sub 2 Te sub 3 et
SOURCE: AN SSSR. Doklady, v. 1	150, no. 5, 1963, 1026-1028
TOPIC TAGS: semiconductors, les	ed telluride, cedmium telluride, bismuth tellur
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TOPIC TAGS: graphite melting point, graphite melting pressure dependence, graphite melting pressure, graphite ABSTRACT: Pressure dependence of the melting point of graphite	E 36994 KX	gpa/gppfal/gap(al/gar(al/BDS APPTG/ARD Paul/Prail ID/Ag
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TOPIC TAGS: graphite melting point, graphite melting pressure dependence, graphite melting pressure, graphite ABSTRACT: Pressure dependence of the melting point of graphite		
ABSTRACT: Pressure dependence of the melting point of graphite		
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FATEYEVA, N.S.; VERESHCHAGIN, L.F.; KOLOTYGIN, V.S.

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Optical method for determining the melting point of graphite as dependent on pressure up to 40,000 atm. Dokl. AN 3SSR 152 no.2:317-319 S '63. (MIRA 16:11)

1. Institut fiziki vysokikh davleniy AN SSSR i Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova. 2. Chlenkorrespondent AN SSSR (for Vereshchagin).

KABALKINA, S.S.; VERESHCHAGIN, L.F.; MYLOV, V.P.

Phase transitions in antimony under high pressure. Dokl. AN SSSR 152 no.3:585-586 S '63. (MIRA 16:12)

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Chler.-korrespondent AN SSSR (for Vereshchagin).

KABALKINA, S.S.; POPOVA, S.V.; SEREBRYANAYA, N.R.; VERESHCHAGIN, L.F.

New modification of Ag₂0 with a laminar structure. Dokl.
AN SSSR 152 no.4;853-854 0 163. (MIRA 16:11)

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Chlen-korrespondent AN SSSR (for Vereshchagin).

SEMERCHAN, A.A.; KUZIN, N.N., DROZDOVA, L.N.; VERESHCHAGIN, L.F.

现的基础,这个人,我们也是一个人的,我们们们的一个人的,我们们们们的一个人的,我们们们们的一个人的一个人的人,我们们们们的一个人的人,我们们们们是一个人的人们们

Variations in the electric resistance of PbS, PbSe, and PbTe at pressures up to 200,000 kg./cm². Dokl. AN SSSR 152 ro.5:1079-1081 0 '63.

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Chlen-korrespondent AN SSSR (for Vereshchagin).

Plastic deformations in a gravitating sphere. Dokl. AN SSSR (MIRA 17:1)

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Chlen-korrespondent AN SSSR (for Vereshchagin).

ACCESSION NE: AP4041747

s/0181/64/006/007/2223/2225

AUTHORS: Vereshchagin, L. F.; Itskevich, Ye. S.; Atabayeva, E. Ya.; Popova, S. V.

TITLE: On a new modification of Bi₂Se₃

SOURCE: Fizika tverdogo tela, v. 6, no. 7, 1964, 2223-2225

TOPIC TAGS: bismuth inorganic compound, polymorphism, metal structure, x ray diffraction study

ABSTRACT: This is a continuation of an earlier study (FTT v. 6, 000, 1964) of the electric resistivity of Bi₂Se₃ as a function of the pressure in the interval up to 140 kbar at room temperature. Along with the previously observed reversible transition to the metallic state observed near 100 kbar at room temperature, an irreversible polymorphic transition was observed at 800C and 120--65 kbar, to

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ACCESSION NR: AP4041747

a new phase Bi2Se3 II which is metastable under normal conditions. To confirm the polymorphic nature of the transition, the sample was annealed for 2 hours in pure helium (500C), and the reverse transition Bi2Se3 II - Bi2Se3 I was established by x-ray diffraction. x-ray diffraction pattern has 40 lines which could be identified in a structure of the bismuth type (Bi2S3), orthorhombic cell, space group Phnm (D2h) The unit cell parameters of the new phase are $a = 11.63 \pm 0.03 \text{ Å, } b = 11.76 \pm 0.03 \text{ Å, and } c = 4.06 \pm 0.01 \text{ Å.}$ density determined by x-ray diffraction and pycnometrically is 7.8 and 8.0 \pm 0.3 g/cm³, respectively, confirming the correctness of the proposed structure. The resistivity of the new phase is 1.2--1.5 ohm-cm, and the temperature coefficient of resistivity is negative between 0 and 100C. The data confirm the correlation between the electric properties and the crystal structure inherent in compounds A2B3 of elements of groups V-VI. Data on the electric properties

ACCESSION NR: AP4041747

of the new phase will be published in the future. "The authors thank S. S. Kabalkina for help with the x-ray diffraction studies." Orig. art. has: 2 tables.

ASSOCIATION: Institut fiziki vy*sokikh davleniy AN SSSR, Moscow (Institute of High Pressure Physics, AN SSSR)

SUBMITTED: 19Mar64

ENCL: 00

SUB CODE: SS

NR REF SOV: 001

OTHER: 002

ACCESSION NR: AP4043610

s/0056/64/047/002/0414/0418

AUTHORS: Vereshchagin, L. F.; Kabalkina, S. S.

TITLE: Phase transitions in antimony at high pressures

SOURCE: Zh. eksper. i teor. fiz., v. 47, no. 2, 1964, 414-418

TOPIC TAGS: high pressure effect, antimony, single crystal, cunic symmetry, phase transition

ABSTRACT: This is a sequel of earlier work by the authors (with V. P. My*lov, DAN SSSR, v. 152, 585, 1963), except that single-crystal antimony was used with A7 structure at room temperature. X-ray diffraction studies have disclosed the presence of two reversible phase transitions: SbI \rightarrow SbII (at 70 kbar into a primitive cubic structure) and SbII \rightarrow SbIII (at 85 kbar -- into a close packed hexagonal structure). The pressure at which the phase transition took place could be determined by plotting the ratio of the

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ACCESSION NR: AP4043610

lattice parameters against the pressure. The phase transitions are accompanied by an increase in the coordination number. In the first transition the coordination number becomes equal to 6 (in place of 3), and in the second transition it becomes equal to 12. The atomic radius assumes in this case values of 1.49 and 1.66 Å, respectively. The results have established that the first transition is the result of gradual removal of distortion from the initial A7 structure. Orig. art. has: 4 figures and 4 tables.

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ASSOCIATION: Institut fiziki vy*sokikh davleniy Akademii nauk SSSR (Institute of High Pressure Physics, Academy of Sciences SSSR)

SUBMITTED: 25Feb64 ENCL: 00

SUB CODE: SS NR REF SOV: 004 OTHER: 007

Card 2/2

ACCESSION NR: AP4012965

s/0020/64/154/004/0819/0820

AUTHOR: Panfilov, V. V.; Vereshchagin, L. F. (Corresponding Member AN SSSR)

TITIE: Paremagnetic resonance in MnS in a wide temperature range

SOURCE: AN SSSR. Doklady*, v. 154, no. 4, 1964, 819-820

TOPIC TAGS: paramagnetic resonance, antiferromagnetic, resonance absorption, mangamese sulfide

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ABSTRACT: The present work was undertaken in order to find the reason for the discrepancy in the results of other authors who studied the resonance of uniaxial antiferromagnetics during the transition from the paramagnetic to the antiferromagnetic state. The resonance in MnS powder has been measured in the temperature interval between +100 to -195C at a frequency of 9285 Mc. The apparatus is described [essentially a double Dewar and heating arrangement]. The constant magnetic field of an electromagnet was perpendicular to the magnetic component of the high-frequency field at the location of the specimen. The resonance absorption maximum at first increases somewhat with decreasing temperature, then decreases rapidly.

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ACCESSION NR: AP4012965

The half-width of the absorption line increases rapidly. The results obtained agree with those of L. R. Maxwell and T. R. McGuire (Rev. Mod. Phys. 25, 279 (1963)). Orig. art. has: 4 figures.

ASSOCIATION: Institut fiziki vy*sokikh davleniy Akademii nauk SSSR (Institute for High-Pressure Physics, Academy of Sciences SSSR)

SUBMITTED: 170ct63

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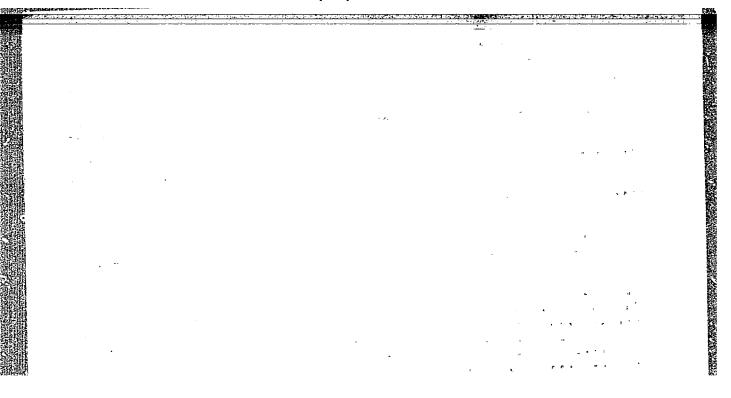
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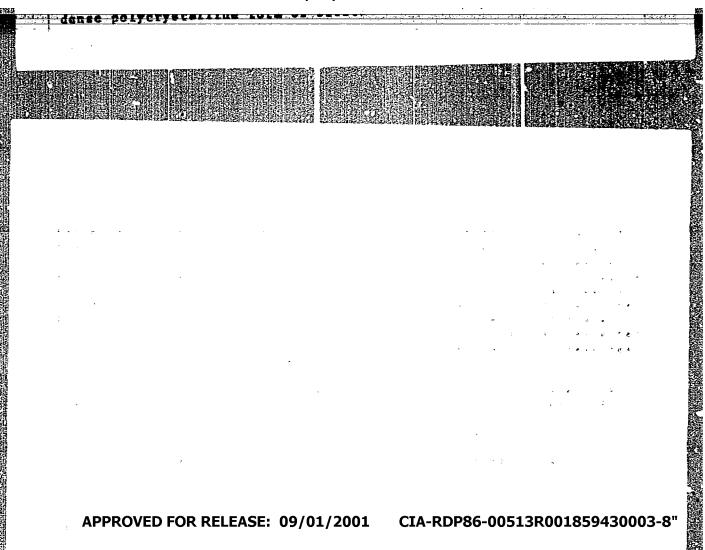
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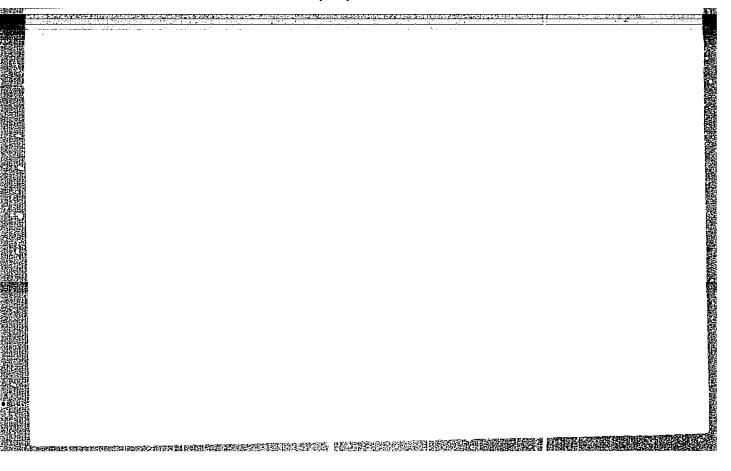
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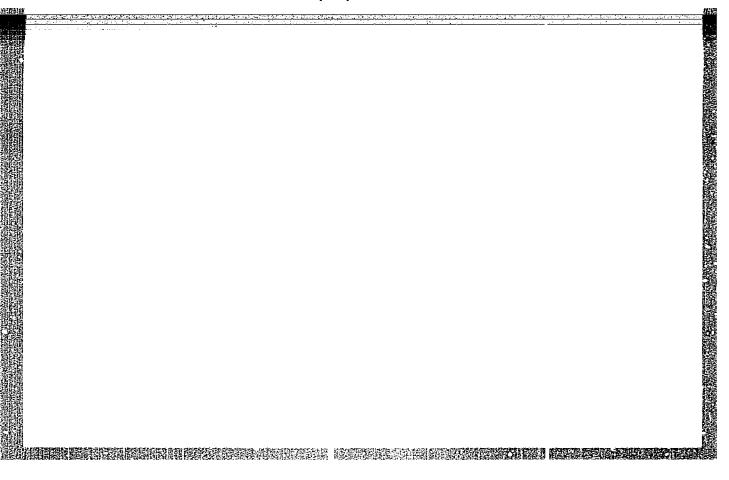
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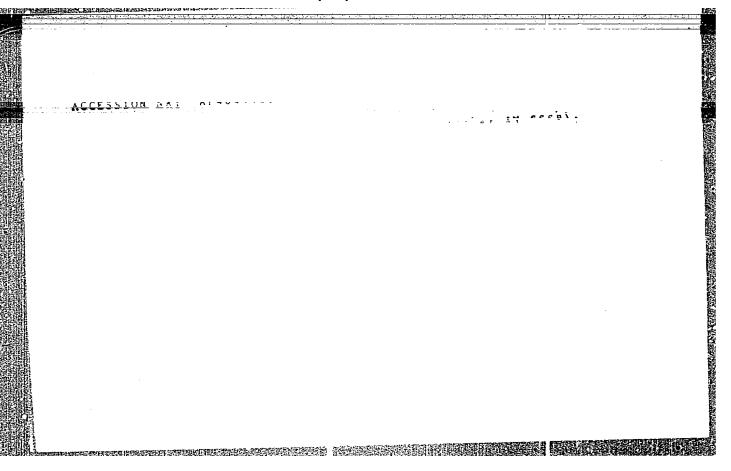


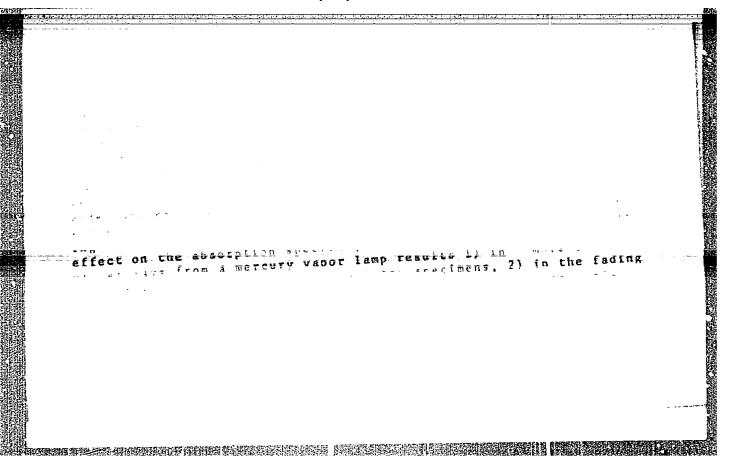


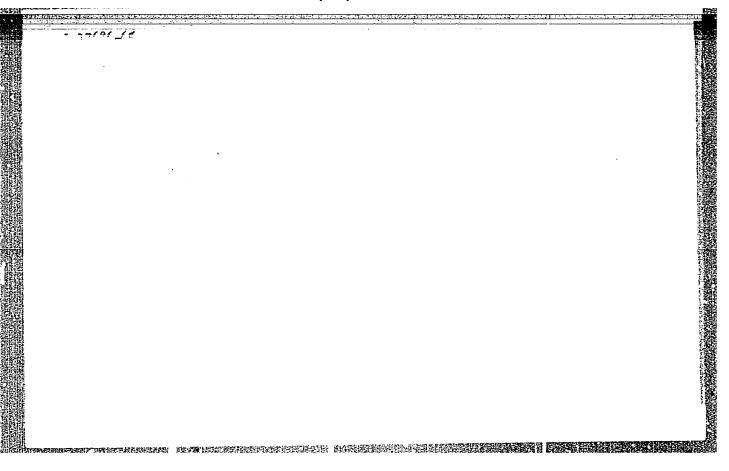
BENDELIANI, N.A.; VERESHCHAGIN, L.F.

Synthesis of dense modifications of silica with the use of water at a pressure of 150·10 kg/cm². Bokl. AN SSR 158 no.4:819-820 0 '64. (MIPA 17:11)

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Chlen-korrespondent AN SSSR (for Vereshchagin).







WERESHCHAGIN, L.F.; IDERCYICH, Ye.J., 21/BAYUTA, E.Ya.; 1007001, E.W.

New modification of Signal Pol. there is a fine Pol. 23 0.225 dl Mac.

(MiR: 17:10)

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VERESHCHAGIN, L.F.; KABALKINA, S.S.

Phase transitions in antimony at high pressures. Zhur. eksp. i teor.

(MIRA 17:10)

fiz. 47 no.2:414-418 Ag 164.

1. Institut fiziki vysokikh davleniy AN SSSR.

VERESHCHAGIN, L.F.; KABALKINA, S.S.; TROITSKAYA, Z.V.

Effect of high pressure on the structure of gallium and indium. Dokl. AN SSSR 158 no.5:1661-1663 0 % 64. (MIRA 17:10)

1. Institut fiziki vysokikh davleniy AN SSSR. 2. Chlen-korrespondent AN SSSR (for Vereshchagin).

Card 1/2

ENT(n)/T/ENP(t)/ENA(h)JD IJP(c) UR/0020/65/163/002/0326/0328, AP5018741 ACCESSION NR: AUTHOR: Vereshchagin, L. F. (Corresponding member AN SSSR); Kabulkina, S. S.; TITIE: Investigation of the influence of high pressure on the structure of tin Lityagina, L. M. SOURCE: AN BESR. Doklady, v. 163, no. 2, 1965, 326-328 oxide TOPIC TAGS: pressure effect, tin compound, crystal lattice structure, phase ABSTRACT: An x-ray investigation of the structure of SnO was made at room temperature and prosures up to 100 kbar. A special x-ray camera (DAN v. 151, no. 5, 1068, 1963; J. Jamieson and A. W. Iawson, J. Appl. Phys. v. 33, no. 3, 776, 1962) with molybdenum radiation was used, the main part of which was a pellet made of amorphous boron and a channel for the sample. The pressure could be determined accurate to 15 kbar. The regults show that at high pressures SnO experiences a reversible phase transition. In most cases this transition occurs at 40--50 kbar, although in was experiments it was observed at 15--20 kbar. The unit cell parameters of the high-pressure phase are a = 3.42 ± 0.02 Å and c = 5.62 ± 0.04 Å. A sudden change in volume of 7.0 ± 5% was observed during the phase transition (at

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ACCESSION NR: AP5018741

10 kbar). The two phases differ from each other in the order of arrangement of the crystal layers and the arrangement of the nearest neighbors of the tin atoms. The results show also that the phase transition is reconstructive, in that the Sn...0 bond in the low-pressure phase is destroyed during the transition and a new bond is produced. It is suggested in analogy with earlier data by others that at higher pressures SnO will experience a polymorph of transition from a wurtzite structure to a NaCl structure. Orig. art. has: 3 f gures and 1 table.

ASSOCIATION: Institut fiziki vysokikh davleniy Akademii nauk SSSR (Institute of

SUBMITTED: 06Apr65

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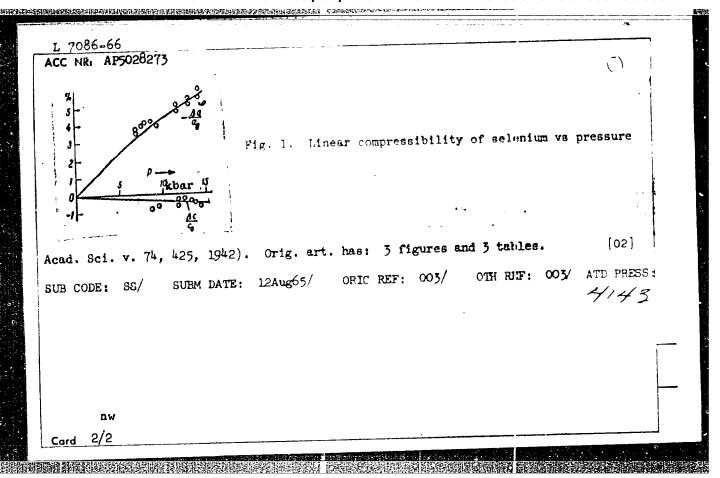
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OTHER: 007

PRG: Institute of Physics of High Pressures, Academy of Sciences, SSSR (Institut Priziki vysokikh davleniy Akademii nauk SSSR) PITTE: X-ray diffraction investigation of the compressibility of hecagonal selenium up to 15 kbar SOURCE: AN SSSR. Doklady, v. 165, no. 2, 1965, 297-298 POPIC TAGS: pressure effect, superhigh pressure, selenium, x ray diffraction study and a selenium and a selenium. The x-ray diffraction study was carried out in a special chamber, described elsewhere (PAN v. 145, 818, 1962), in which the high-pressure vessel was a cone of metallic beryllium with a channel (1.500 measured with a manganin manometer accurate to fICC bar. The hexagonal selenium modification was prepared from the amorphous one at the kbar at 400°. The results (Fig. 1) show that selenium has a highly anisotropic compressibility, similar to that of tel-	JUNEOD, Vereshohagin, L. F. (Correspondi	ng member AN SSSR); Kabalkira, S. S.; Shule-
TITIE: X-ray diffraction investigation of the compressibility of hecegonal selenium up to 15 kbar BOURCE: AN SSSR. Doklady, v. 165, no. 2, 1965, 297-298 TOPIC TAGS: pressure effect, superhigh pressure, selenium, x ray diffraction study (Zhett v. 45, 2073, 1963) are extended to include hexagonal selenium. The x-ray diffraction study was carried out in a special chamber, described elsewhere (2AN v. 143, 818, 1962), in which the high-pressure vessel was a cone of metallic beryllium with a channel (1. mm which the sample. Aviation gasoline was used to transmit the pressure, which was measured with a manganin manometer accurate to t1(C bar. The hexagonal selenium modification was prepared from the amorphous one at the kbar at 400°. The results (Fig. 1) show that selenium has a highly anisotropic compressibility, similar to that of tel-	nin, B. M.	
TITIE: X-ray diffraction investigation of the compressibility of hecagonal selenium up to 15 kbar BOURCE: AN SSSR. Doklady, v. 165, no. 2, 1965, 297-298 TOPIC TAGS: pressure effect, superhigh pressure, selenium, x ray diffraction study ABSTRACT: Earlier studies by the authors on single-crystal tellurium (ZhETF v. 45, 2073, 1963) are extended to include hexagonal selenium. The x-ray diffraction study was carried out in a special chamber, described elsewhere (2AN v. 143, 818, 1962), in which the high-pressure vessel was a cone f metallic beryllium with a channel (m which the sample. Aviation gasoline was used to transmit the pressure, which was measured with a manganin manometer accurate to t1(C bar. The hexagonal selenium modification was prepared from the amorphous one at tx kbar at 400°. The results (Fig. 1 show that selenium has a highly anisotropic compressibility, similar to that of tel-	fiziki vysokikh davleniy Akademii nauk bo	ÖK;
ABSTRACT: Earlier studies by the authors on single-crystal tellurium (ZhETF v. 45, 2073, 1963) are extended to include hexagonal selenium. The x-ray diffraction study was carried out in a special chamber, described elsewhere (DAN v. 143, 818, 1962), in which the high-pressure vessel was a cone of metallic beryllium with a channel (1. mm diam) for the sample. Aviation gasoline was used to transmit the pressure, which was measured with a manganin manometer accurate to t100 bar. The hexagonal selenium modification was prepared from the amorphous one at the kar at 400°. The results (Fig. 1 show that selenium has a highly anisotropic compressibility, similar to that of tel-	TITLE: X-ray diffraction investigation	of the compressibility of hecagonal selenium
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ABSTRACT: Earlier studies by the authors on single-crystal tellurium (ZhETF v. 45, 2073, 1963) are extended to include hexagonal selenium. The x-ray diffraction study was carried out in a special chamber, described elsewhere (DAN v. 143, 818, 1962), in which the high-pressure vessel was a cone of metallic beryllium with a channel (1.000 m) which the sample. Aviation gasoline was used to transmit the pressure, which was measured with a manganin manometer accurate to t100 bar. The hexagonal selenium modification was prepared from the amorphous one at the kbar at 400°. The results (Fig. 1 show that selenium has a highly anisotropic compressibility, similar to that of tel-	TOPIC TAGS: pressure effect, superhigh I	ressure, selenium, x ray diffraction study
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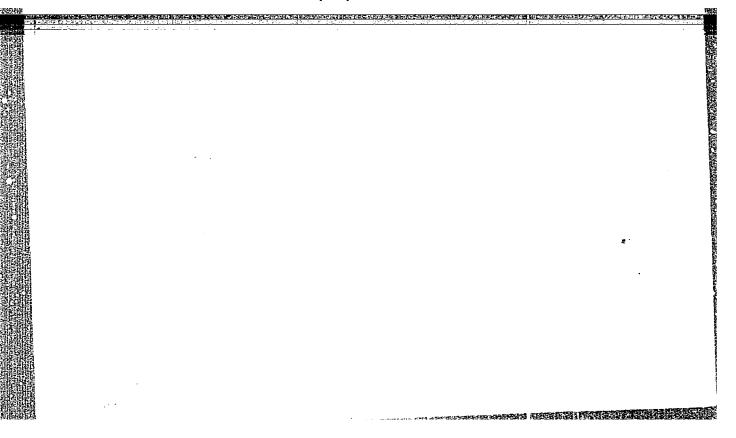
ARTSIMOVICH, L.A., akademik; KELDYSH, M.V., akademik; KAPITSA, P.L., akademik; WIL, B.M.; VERESHCHAGIN, L.F.; PISTOL'KORS, A.A.; SHCHUKIN, A.N., akademik; SKOBELITSYN, D.V., akademik; ALEKSANDROV, A.P., akademik; AMBARTSUMYAN, V.A., akademik; ZEL'DOVICH, Ya.B.; SEMEROV, N.N., akademik; KOTEL'NIKOV, V.A., akademik; LIFSHITS, 1.M.; VLKSHER, V.I., akademik; GINZBURG, V.L.; MILLIONSHCHIKOV, N.D., akademik

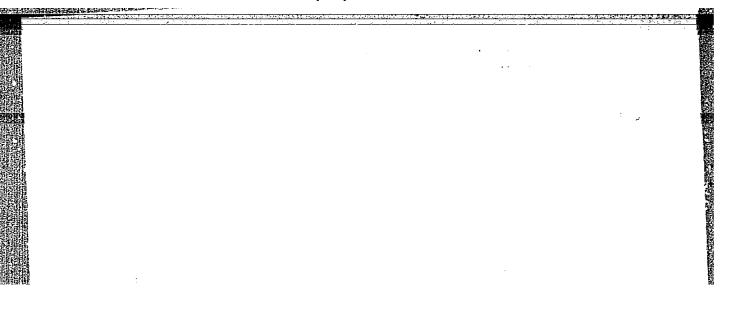
Some problems in the development of modern physics; discussion of the work of the Department of General and Applied Physics. Vest. AN SSSR 35 no.2:3-46 F 165. (MIRA 18:3)

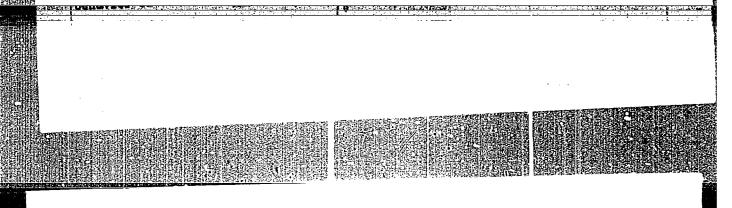
1. Chleny-korrespondenty AN SSSR (for Vul, Vereshchagin, Pistol'kors, Lifshits, Ginzburg).

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FEDYUSHIN, N.D.; DIKUSHIN, V.I., akademik, retsenzent; VERESHOHAGIN, L.F., retsenzent; SUVORINA, L.N., inzh., red.

[Selecting optimal variants of thick-walled structures; hardbook] Vybor optimal'nykh variantov tolstostennykh konstruktsii; spravochnoe posobie. Moskva, Mashinostroenie, 1965. 81 p. (MIRA 18:5)

1. Chlen-korrespondent AN SSSR (for Vereshchagin).

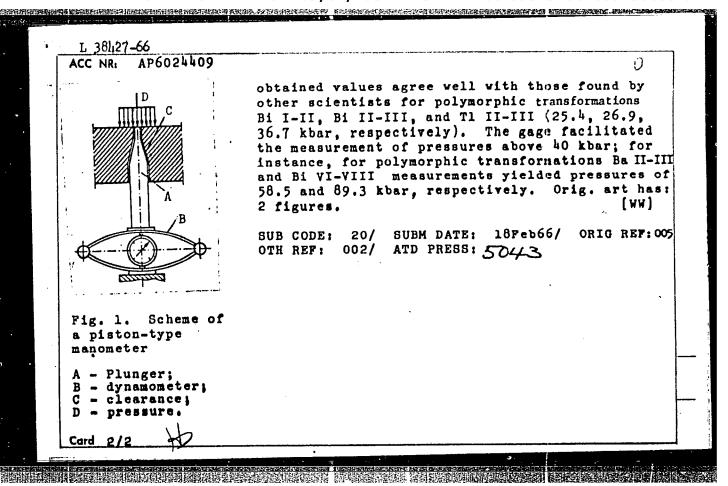
VERESHCHAGIN, L.F., KABALKINA, G.G., ECH SETT, A.F.

Phase transition in MrF₂ at high pressure. Hur.cksp. i teor. fir. 19 no.0s1728.1732 D 165. (Nice 19:1)

1. Institut ficiki vysokikh davleniy AM 1.39. Selectted July 12,

EWT(m)/EWP(k)/EWP(t)/ETI JP(c) HW/JD L 38127-66 UR/0020/66/169/001/0074/0076 SOURCE CODE: ACC NR: AP6024409 AUTHOR: Vereshchagin, L. F. (Corresponding member Ad SSSR); Zubova, Burdina, K. P.; Buymova, I. P. ORG: Institute of High Pressures, Academy of Sciences SSSR (Institut vysokikh davleniy Akademii nauk SSSR) Measuring pressures up to 100 kbar/by the free-plunger method TITLE: v. 169, no. 1, 1966, 74-76 SOURCE: AN SSSR. TOPIC TAGS: pressure, high pressure, pressure measurement, high pressure measurement, pressure gage, high pressure research, metal test polymerutic lang towns then ABSTRACT: A pressure gage based on the free-plunger principle, for measuring pressures up to 100 kbar, has been designed and built. The pressure in the high-pressure chamber is measured directly by a spring dynamometer connected to a free plunger (see Fig. 1). The friction of the plunger is reduced to an insignificant value by the special configuration of the plunger, and by a special lubricant filling clearance C. The top and bottom ends of the plungerare coaxial cylinders, which ensures the stability of the plunger and prevents a runoff of the compressed substance. The pressure gage was used for measuring the pressure of the polymorphic transformation of somes metals. UDC: 539.89 Card 1/2

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	ACC NR: AP6037065 (A) SOURCE CODE: UR/0030700703170031
	AUTHOR: Kabalkina, S. S.; Vereshchagin, L. F.; Serebryanaya, N. R.
	ORG: Institute of Physics of High Pressures, Academy of Sciences, SSSR (Institut 1121R) vysokykh davleniy Akademii nauk SSSR)
	TITLE: Germanium telluride phase transformation under high pressure
	council aburnal eksperimental noy i teoreticheskoy fiziki, v. 51, no. 5, 1966,
	1358-1362 crystal place from the form of t
	phase transformation, pressure error
	ABSTRACT: The effect of high pressures up to 100 kbar on the crystal structure of GeTe has been investigated. A phase transition from a rhombohedral phase GeTeI (GeTe has observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-ray diffraction (A7-type) to a cubic phase of GeTeII (NaCl type) was observed. X-
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UR/0020/67/172/001/0076/0076 SOURCE CODE: ACC NR: AP7003645 AUTHOR: Kalashnikov, Ya.A.; Feklichev, Yc.N.; Sukhushina, I.S.; Vereshchagin, L.F. (Academician) ORG: Institute of Physics of High Pressures, Academy of Sciences, SSSR (Institut fiziki vyšokikh davleniy Akademii nauk SSSR); Moscow State University (Moskovskiy gosudarstvennyy universitet) Production of ballas-type synthetic diamonds TITLE: SOURCE: AN SSSR. Doklady, v. 172, no. 1, 1967, 76 and insert facing p. 76 TOPIC TAGS: synthetre diamond, synthetic diamond production for the tro ABSTRACT: Synthetic diamonds up to 6-6.5 mm in size with a central-zone density diamond, structure cupital higher than that of natural diamonds have been produced. The density decreases to standard level at the specimen surface, which consisted of fine bound crystals. The internal and surface structure of the synthetic diamonds compared very closely to the ballas structure of natural diamond. 11, 13/ SUBM DATE: 24Sep66/ ORIG REF: 001/ OTH REF: 006/ SUB CODE: ATD PRESS: 5114 UDC:_ 666.233

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SOURCE CODE: UR/0020/67/172/002/0313/0316

AUTHOR: Kabalkina, S. S.; Kolobyanina, T. N.; Vereshchagin, L. F. (Academician)

ORG: Institute of High Pressure Physics, Academy of Sciences, SSSR (Institut fiziki

vysokikh davleniy Akademii nauk SSSR)

TITIE: X ray diffraction investigation of the crystal structure of iodine at high pressure

SOURCE: AN SSSR. Doklady, v. 172, 2, 1967, 313-316

TOPIC TAGS: x ray diffraction study, iodine, high pressure research, crystal lattice structure, molecular crystal

ABSTRACT: The tests on iodine were made because at high pressure it is one of the few elements having a molecular structure, and may be the only element in which structure investigations can be made at room temperature. The authors carried out an x-ray diffraction study of its structure at room temperature and pressures up to 60 ktar, using a procedure described earlier (DAN v. 151, 1068, 1963) and molybdenum radiation. To improve the diffraction pattern, the iodine was tested in powdered form. The observed appearance and disappearance of several lines is reported, as well as coalescence of some lines with variation of pressure. In addition, the pressure dependence of the volume of the iodine and of the parameters of its lattice structure are plotted. The results indicate that pressure does not change the initial rhombic structure, merely distorting it and leading to some rotatior of the molecules. It is

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GILLER, S.A., akademik; VERESHCHAGIN, L.I.; VENTER, K.K.; KOPSHUNOV, S.P.; TSIRULE, V.V. [Cirule, V.]; LOLYA, D.O.

在制度的**,但是对于自己的,**

2-Furyl and 5-nitrofuryl-2-acetylene ketones. Dokl. AN SSSR 164 no.1:99-102 S '65. (MIRA 18:9)

l. Institut organicheskogo sinteza AN Lr'viyskoy SSR i Institut nefte- i uglekhimicheskogo sinteza pri Irkutskom gosudarstvennom universitete im. A.A. Zhdanova. 2. An Latviyskoy SSR (for Giller).